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Microstructural evolution of new type Al–Zn–Mg–Cu alloy with Er and Zr additions during homogenization

Hao WU, Sheng-ping WEN, Jun-tai LU, Zhen-peng MI, Xian-long ZENG, Hui HUANG, Zuo-ren NIE

School of Materials Science and Engineering, Beijing University of Technology, Beijing 100124, China

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Abstract: A comprehensive study on the microstructural evolution of a new type Al–Zn–Mg–Cu–Er–Zr alloy during homogenization was conducted by optical microscope, scanning electron microscope, transmission electron microscopy and X-ray diffraction analysis. The results show that serious segregation exists in as-cast alloy, and the primary phases are T(AlZnMgCu), S(Al₂CuMg) and Al₈Cu₄Er, which preferentially locate in the grain boundary regions. The soluble T(AlZnMgCu) and S(Al₂CuMg) phases dissolve into the matrix gradually during single-stage homogenized at 465 °C with prolonging holding time, but the residual Al₈Cu₄Er phase cannot dissolve completely. Compared with the single-stage homogenization, both a finer particle size and a higher volume fraction of L1₂-structured Al₃(Er,Zr) dispersoids can be obtained in the two-stage homogenization process. A suitable homogenization scheme for the present alloy is (400 °C, 10 h)+(465 °C, 24 h), which is consistent with the results of homogenization kinetic analysis.

Key words: Al-Zn-Mg-Cu-Er-Zr alloy; homogenization; microstructural evolution; primary phases; Al₃(Er,Zr) particles

1 Introduction

The comprehensive performance improvement is an eternal pursuit in the development of aluminum alloy. High-strength 7xxx series Al-Zn-Mg-Cu alloys have been widely used for structure components in aircraft and aerospace industries due to their outstanding properties, such as high strength, fracture toughness and stress corrosion cracking resistance [1,2]. Over the past years, intensive researches have been conducted to improve the comprehensive performance of Al-Zn-Mg-Cu alloys via various methods [3]. WU et al [4,5] have carried out in-depth investigation on erbiumcontaining aluminium alloy and developed a new type Al-Zn-Mg-Cu alloy microalloyed with Er and Zr elements. The new type Al-Zn-Mg-Cu alloy possesses an outstanding comprehensive performance due to the formation of stable and coherent L12-structured Al₃(Er,Zr) dispersoids, which can inhibit recrystallization by grain boundary pinning during subsequent hightemperature processing [6,7].

As well known, homogenization treatment, as an

indispensable processing step, plays a key role in removing the microsegregation and dissolving large soluble non-equilibrium intermetallic phases formed during solidification [8-11]. In addition, a suitable homogenization scheme is crucial for the formation of fine dispersoid particles [12–14]. A large amount of work has been done on Al-Zn-Mg-Cu alloys homogenization [15,16]. LI et al [17] investigated the microstructural evolution of Al-Zn-Mg-Zr alloy with trace amount of Sc during homogenization treatment. SHI et al [18] found that the evolution of primary eutectic structure of 7085 alloy during homogenization consists of three processes: dissolution of eutectic $\alpha(Al)+T(AlZnMgCu)$ microstructure, phase transformation from T to $S(Al_2CuMg)$ phase and the dissolution of S phase. LÜ et al [14] studied the distribution of Al₃Zr dispersoids in commercial 7150 alloy, and concluded that two-step homogenization treatments resulted in Al₃Zr dispersoids with a finer particle size and a larger number density and volume fraction. Moreover, ROBSON [19] verified that a two-step homogenization treatment can improve the dispersoid particle distribution and minimize the width of precipitation free zone in the region near the grain

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Corresponding author: Hao WU; Tel/Fax: +86-10-67391536; E-mail: wuhao2469@126.com DOI: 10.1016/S1003-6326(17)60168-7

boundary, which results in a significant decrease in the recrystallized fraction.

In this study, in order to eliminate the large soluble non-equilibrium intermetallic phases and obtain a homogeneous distribution of $L1_2$ -structured $Al_3(Er,Zr)$ dispersoids, a series homogenization schemes of the present as-cast Al-Zn-Mg-Cu-Er-Zr alloy have been conducted. Meanwhile, the microstructural evolution of the new type Al-Zn-Mg-Cu-Er-Zr alloy during homogenization treatments was investigated.

2 Experimental

The as-cast Al-Zn-Mg-Er-Zr alloy used in this study was provided by Southwest Aluminum Co., Ltd., China. The chemical composition (mass fraction, %) of the Al-Zn-Mg-Cu-Er-Zr alloy used in this study is as follows: 5.6 Zn, 2.1 Mg, 1.2 Cu, 0.1 Er, 0.1 Zr, 0.1 Mn, 0.07 Fe, 0.1 Si, and balance Al. Specimens with dimensions of 10 mm \times 10 mm \times 7 mm were used for homogenization treatments. Seven types of homogenization conditions were conducted as below: (465 °C, 6 h), (465 °C, 12 h), (465 °C, 24 h), (465 °C, 48 h), (465 °C, 72 h), (400 °C, 10 h) and (400 °C, 10 h)+ (465 °C, 24 h). The microstructural evolution of the as-cast and homogenized samples was characterized using Olympus BX51M optical microscope (OM), FEI Quanta 650 FEG scanning electron microscope (SEM) and JEOL 2100 transmission electron microscopy (TEM) with an operating voltage of 200 kV. The OM specimens were polished and etched with the Keller's reagent solution at room temperature. SEM observation was carried out in the backscattered electron imaging mode, operated at 20 kV. The map scanning analyses were conducted on an energy dispersive X-ray spectrometer operated at 20 kV. X-ray diffraction (XRD) studies were performed on a D/max 2500PC diffractometer with Cu K_{a1} radiation. TEM foils were prepared by mechanical polishing to less than 100 µm, subsequently punched into 3 mm discs and final twin-jet polishing with an electrolyte solution of 30% nitric acid and 70% methanol at the voltage of ~15 V DC and the temperature below -25 °C.

3 Results and discussion

3.1 Characterization of as-cast microstructure

The OM and SEM microstructures of the as-cast Al-Zn-Mg-Cu-Er-Zr alloy are shown in Fig. 1. It can be seen that a large amount of coarse phases distribute along grain boundaries. Figure 1(b) shows a higher magnification image of the coarse phase, which is similar to the lamellar eutectic structure. Three zones with different composition contrasts, as shown by arrows A, Band C, were identified by energy dispersive X-ray spectrometry (EDX), and the EDX results are listed in Table 1. EDX analysis reveals that the light-grey (as indicated by A), the dark grey (B) and the bright white (C)contrast intermetallic phases can be identified as T(AlZnMgCu), $S(Al_2CuMg)$ and Al_8Cu_4Er type compounds, respectively. Figure 2 shows the scanning electron microstructure and the corresponding elements of Zn, Mg, Cu, Er and Zr distribution in the as-cast alloy. The main elements Zn, Mg and Cu and the microalloying



Fig. 1 OM (a) and SEM (b) images of as-cast Al-Zn-Mg-Cu-Er-Zr alloy

Table 1 EDX results of intermetallic phases in as-cast	Al–Zn–Mg–Cu–Er–Zr alloy (mole fraction, %)
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Zone	Al	Zn	Mg	Cu	Mn	Er	Zr	Fe	Si
A	38.81	24.81	27.97	8.28	0.02	0.01	0.00	0.08	0.01
В	63.73	2.92	17.97	15.08	0.10	0.04	0.04	0.03	0.08
С	60.33	4.69	0.93	25.25	0.12	6.02	0.18	2.48	0.01



Fig. 2 SEM images and corresponding elements distribution maps of as-cast alloy

element Er are enriched at grain boundaries. Therefore, a proper homogenization treatment is required to remove severe microsegregation in as-cast Al–Zn–Mg–Cu–Er–Zr alloy.

3.2 Microstructural evolution during homogenization

Figure 3 shows the evolution of statistical amount of the grain boundary phase after homogenization at 465 °C for various time. The statistics of every condition was analyzed using Image-Pro Plus software and by selecting more than 10 images at a same magnification. The grain boundary phase firstly decreased sharply within the initial 12 h, and its amount decreased slightly with further prolonging of homogenization time.

Figure 4 displays the backscattered electron images of the residual phases under various homogenization conditions, and the corresponding EDX results of the intermetallic phases are shown in Table 2. After homogenizing at 400 °C for 10 h, three zones with different compositions contrast still exist (as arrowed A, B and C in Fig. 4(a)). According to the EDX results, the three zones A, B and C are T(AlZnMgCu), S(Al₂CuMg) and Al₈Cu₄Er phases, respectively. The



Fig. 3 Evolution of statistical amount of grain boundary phase after homogenizing at 465 °C for various time

content of Cu in T(AlZnMgCu) phase in this condition is a little higher than that of T(AlZnMgCu) phase in as-cast alloy, which could be attributed to the lower diffusivity of Cu. The type of residual phase in the alloy after homogenizing at 465 °C for 6 h, as arrowed by D, E and F in Fig. 4(b), is roughly similar to that in the as-cast alloy and alloy homogenized at 400 °C for 10 h. But the continuous lamellar structures dissolved gradually and



Fig. 4 Backscattered electron images of alloys under different homogenization conditions: (a) 400 °C, 10 h; (b) 465 °C, 6 h; (c) 465 °C, 12 h; (d) 465 °C, 24 h; (e) 465 °C, 48 h; (f) (400 °C, 10 h) + (465 °C, 24 h)

Position	Al	Zn	Mg	Cu	Mn	Er	Zr	Fe	Si
A	33.59	28.43	20.49	17.30	0.00	0.00	0.00	0.08	0.11
В	57.93	4.25	16.03	21.59	0.00	0.00	0.00	0.00	0.19
С	49.76	12.66	3.43	27.88	0.10	4.59	0.02	1.60	0.06
D	53.04	13.80	21.37	11.61	0.03	0.01	0.00	0.02	0.11
Ε	51.92	1.81	24.69	21.34	0.03	0.00	0.03	0.06	0.11
F	62.24	3.61	0.79	23.36	0.23	6.13	0.00	3.58	0.07
G	62.78	4.03	0.32	23.04	0.11	6.30	0.00	3.29	0.13
H	63.67	4.06	0.63	22.65	0.19	5.85	0.00	2.88	0.08
Ι	62.65	4.01	0.42	21.70	0.31	6.33	0.00	4.44	0.14
J	62.70	4.12	0.30	23.56	0.13	6.38	0.02	2.79	0.00

Table 2 EDX results of intermetallic phases in Fig. 4 (mole fraction, %)

evolved to isolate particles after homogenization at 465 °C. With further prolonging of homogenization time, the *T* and *S* phases gradually dissolve into matrix and disappear finally. The residual bright white contrast phases, as arrowed in Figs. 4(c)-(f), are identified as the Al₈Cu₄Er phase by EDX analysis, which cannot be dissolved and eliminated completely in Al matrix during homogenization treatment at 465 °C for 48 h. This is due to the fact that its melting point is higher than the homogenization temperature.

X-ray diffraction (XRD) patterns of the as-cast and homogenized alloys are shown in Fig. 5. The as-cast alloys are mainly composed of $\alpha(Al)$, $\eta(MgZn_2)$, S(Al₂CuMg) phase. Compared with XRD patterns of the as-cast alloy, there is no obvious difference in the peaks of S(Al₂CuMg) phase in alloy homogenized at 400 °C for 10 h. The $S(Al_2CuMg)$ phase, which formed during the non-equilibrium solidification process, has not dissolved into α (Al) matrix after 400 °C, 10 h pre-treatment. With prolonging time of homogenization treatment at 465 °C for 24 h, the peak of S(Al₂CuMg) phase disappears. The Al₈Cu₄Er phase cannot be detected by XRD, which may be attributed to its low content in the experimental alloys. Moreover, except for $\alpha(Al)$, there is no obvious diffraction peak in XRD patterns after being homogenized at 465 °C for 48 h and (400 °C, 10 h) + (465 °C, 24 h).

Figure 6 exhibits TEM images of the as-cast and homogenized alloys. Only rod-shape non-equilibrium



Fig. 5 XRD patterns of as-cast and homogenized alloys

T(AlZnMgCu) phase exists in the as-cast alloy, as shown in Fig. 6(a). No Al₃(Er,Zr) particles can be detected in this condition. The non-equilibrium *T*(AlZnMgCu) phase disappears and homogeneously distributed Al₃(Er,Zr) precipitates form after homogenization at 465 °C for 24 h. It can be seen from Fig. 6(c) that both finer particle size and higher number density of Al₃(Er,Zr) dispersoids are obtained after homogenization in (400 °C, 10 h)+ (465 °C, 24 h) condition, because the first step held at 400 °C has an significant effect on the precipitation of Al₃(Er,Zr) particles. Figure 6(d) shows the higher magnification bright field TEM image of Fig. 6(c), and the inset selected area diffraction (SAD) pattern implies



Fig. 6 TEM images of as-cast and homogenized alloys: (a) As-cast; (b) 465 °C, 24 h; (c, d) (400 °C, 10 h) + (465 °C, 24 h)

that the Al₃(Er,Zr) precipitates have the Ll₂ structured lattice, as shown in the upper right inset of Fig. 6(d). The first step homogenized at 400 °C for 10 h can promote the nucleation of a large number of dispersoids out of the Al matrix due to a high Er and Zr supersaturation at low temperature. Therefore, much finer particle size and more homogenous distribution of Al₃(Er,Zr) dispersoids can be obtained in (400 °C, 10 h) + (465 °C, 24 h) condition.

3.3 Homogenization kinetic analysis

Generally, homogenization time is determined by the interdendritic phase spacing. Effect of the spacing on the homogenization process could be discussed by the homogenization kinetic analysis. According to the theory of SHEWMAN [20], the homogenization kinetic equation is mathematically expressed as follows:

$$\frac{1}{T} = A \ln \left(\frac{t}{BL^2} \right) \tag{1}$$

where A=R/Q, $B=4.6/(4\pi^2 D_0)$, *R* is the mole gas constant; *Q* is the diffusion activation energy; D_0 is the diffusion coefficient; *L* is the interdendritic *T*(AlZnMgCu) phase spacing; *T* ant *t* are the homogenization temperature K and time s, respectively.

If the parameters of as-cast microstructure are given, the homogenization kinetic curves can be obtained. The diffusion coefficient of Cu in Al is much lower than Mg, and Zn at the same temperature [21]. So, the homogenization process is mainly controlled by the diffusion process of Cu [22-24]. SEM analysis shows that the average interdendritic phase spacings (L) in the studied alloy after the first step and the second step homogenization treatment are 60 and 32 µm, respectively. By substituting of $D_{0(Cu)}=0.084 \text{ cm}^2/\text{s}$, $Q_{(Cu)} =$ 136.8 kJ/mol and R=8.31 kJ/(mol·K) into Eq. (1), the homogenization kinetic curves of Al-Zn-Mg-Cu-Er-Zr alloy for a certain interdendritic phase spacing can be obtained, as shown in Fig. 7. This reveals that the



Fig. 7 Curves of homogenization kinetics at interdendritic phase spacing of $60 \ \mu m$

holding time increases dramatically with the decrease of the homogenization temperature. According to the homogenization kinetic curve, the optimized homogenization conditions of the present as-cast Al–Zn–Mg–Cu–Zr–Er alloy are 25.2 h and 465 °C, which are well consistent with the experimental results. Therefore, taking the precipitation of Al₃(Er,Zr) particles into comprehensive consideration, the most optimized homogenization scheme for the present Al–Zn–Mg–Cu– Zr–Er alloy is (400 °C, 10 h) + (465 °C, 24 h).

4 Conclusions

1) Serious segregation exists in the as-cast alloy, and the primary phases preferentially locate in the grain boundary regions, which are mainly T(AlZnMgCu), $S(Al_2CuMg)$ and Al_8Cu_4Er phases.

2) T(AlZnMgCu) and S(Al₂CuMg) phases dissolve completely into Al matrix after homogenization at 465 °C for 24 h, but the Al₈Cu₄Er phase cannot be dissolved and eliminated in Al matrix because its melting point is higher than the homogenization temperature.

3) Compared with the single-stage homogenization treatment, both a finer particle size and a higher number density of $Al_3(Er,Zr)$ dispersoids can be obtained with the two-stage homogenization treatment.

4) Taking the precipitation of $Al_3(Er,Zr)$ particles into consideration, the optimized homogenization scheme for the present Al–Zn–Mg–Cu–Zr–Er alloy is (400 °C, 10 h) + (465 °C, 24 h), which is consistent with the results of homogenizing kinetic analysis.

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新型 Er、Zr 微合金化 Al-Zn-Mg-Cu 合金 在均匀化中的显微组织演变

吴 浩, 文胜平, 卢军太, 米振鹏, 曾宪龙, 黄 晖, 聂祚仁

北京工业大学 材料科学与工程学院,北京 100124

摘 要:通过使用光学显微镜、扫描电镜、透射电镜及 X 射线衍射等手段研究一种新型 Er、Zr 微合金化 Al-Zn-Mg-Cu 合金在均匀化过程中的显微组织演变。结果表明:铸态合金组织存在严重的偏析,此时合金中含 有大量的 T(AlZnMgCu)、S(Al₂CuMg) 和 Al₈Cu₄Er 相,这些初生相大量偏聚于晶界。随着在 465 °C 单级均匀化 处理的进行,第二相含量大幅度降低,可熔的 T 相和 S 相会逐步地熔入基体,但 Al₈Cu₄Er 相不能完全消除,仍有 少量残留。相对于单级均匀化工艺,合金在双级均匀化处理时不仅能够消除铸态合金偏析组织,而且能够析出大 量细小弥散分布的 L1₂ 结构的 Al₃(Er,Zr)相。结合均匀化动力学分析,可以得出合金合理的均匀化热处理制度为 (400 °C, 10 h) + (465 °C, 24 h)。

关键词: Al-Zn-Mg-Cu-Er-Zr 合金; 均匀化; 显微组织演变; 初生相; Al₃(Er,Zr)粒子

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1482