

Available online at www.sciencedirect.com



Trans. Nonferrous Met. Soc. China 20(2010) s231-s235

Transactions of Nonferrous Metals Society of China

www.tnmsc.cn

## Preparation of YVO<sub>4</sub>:RE (RE=Yb<sup>3+</sup>/Er<sup>3+</sup>, Yb<sup>3+</sup>/Tm<sup>3+</sup>) nanoparticles via microemulsion-mediated hydrothermal method

ZHANG Jun(张 军)<sup>1</sup>, Wulantuya(乌兰图雅)<sup>1</sup>, DI Xiao-wei(狄晓威)<sup>1</sup>, LIU Zhi-liang(刘志亮)<sup>1</sup>, XU Gang(徐 刚)<sup>2</sup>, XU Sheng-ming(徐盛明)<sup>2</sup>

College of Chemistry and Chemical Engineering, Inner Mongolia University, Hohhot 010021, China;
Institute of Nuclear and New Energy Technology, Tsinghua University, Beijing 100084, China

Received 6 July 2009; accepted 30 December 2009

Abstract: A microemulsion-mediated hydrothermal method for synthesis of  $YVO_4$ :RE (RE=Yb<sup>3+</sup>/Er<sup>3+</sup>, Yb<sup>3+</sup>/Tm<sup>3+</sup>) nanoparticles by hydrothermal treatment of quaternary microemulsion medium consisting of Na<sub>3</sub>VO<sub>4</sub>/NaOH and RE(NO<sub>3</sub>)<sub>3</sub> aqueous solution , surfactant cetyltrimethylammonium bromide (CTAB), cosurfactant *n*-hexanol and oil phase *n*-heptane was report. The confinement of microemulsion droplets acting as microreactors during the reaction process allows the formation of small size YVO<sub>4</sub>:RE nanoparticles with relatively narrow size distribution and less aggregation. The structure, size and shape of YVO<sub>4</sub>:RE nanoparticles were investigated by means of X-ray diffractometry (XRD) and transmission electron microscopy (TEM). Compared with the conventional solid annealing diffusion method, the microemulsion-mediated hydrothermal method shows superiority in obtaining YVO<sub>4</sub>:RE nanoparticles with controllable size, narrow size distribution and less aggregation. The microemulsion-mediated hydrothermal method may be potentially applicable for synthesis of other rare earth doped up-converting luminescence nanomaterials.

Key words: yttrium orthovanadate; doping; up-conversion luminescence; microemulsion; hydrothermal method

## **1** Introduction

Since the first discovery of up-conversion luminescence phenomenon, up-converting luminescent materials have been widely investigated due to their unique up-conversion luminescent properties that may find potential applications in several new technologies, such as solid lighting, displaying, biological assays and imaging[1–2]. Many series of materials systems showing up-converting luminescent properties have been developed, and it was recently found that novel and efficient up-conversion luminescence could be realized in nanoscaled up-converting materials[3–6].

Among the widely investigated up-converting nanomaterials systems, up-conversion luminescent yttrium orthovanadate (YVO<sub>4</sub>) nanoparticles doped with rare-earth ions have attracted much research attention in recent years[7–8]. Rare earth ions of  $Yb^{3+}/Er^{3+}$  and  $Yb^{3+}/Tm^{3+}$  are believed to be the most effective ions for doping into  $YVO_4$  host lattice to result in efficient

up-conversion luminescence. However, most of these reports so far are mainly focused on the study of  $YVO_4$  nanomaterials doped with one kind of rare earth ion, such as  $Eu^{3+}$ ,  $Tm^{3+}$ ,  $Dy^{3+}$ ,  $Sm^{3+}$  or  $Er^{3+}[9-11]$ . The reports on co-doping of  $Yb^{3+}/Er^{3+}$  and  $Yb^{3+}/Tm^{3+}$  in  $YVO_4$  nanomaterials are still far more adequate, and the effects of co-doping  $Yb^{3+}/Er^{3+}$  and  $Yb^{3+}/Tm^{3+}$  into  $YVO_4$  nanomaterials resulting in the enhanced up-converting luminescence still need to be further studied.

We have developed a microemulsion-mediated hydrothermal method available for preparation of luminescent nanoparticles with controllable sizes and narrow distribution[12–13]. The method can also be used for preparation of undoped and rare earth doped YVO<sub>4</sub> nanoparticles[13]. The sizes of the nanoparticles could be adjusted by modulating the microemulsion compositions and reaction conditions. In this work, we reported the synthesis of YVO<sub>4</sub>:RE (RE=Yb<sup>3+</sup>/Er<sup>3+</sup>, Yb<sup>3+</sup>/Tm<sup>3+</sup>) nanoparticles via this microemulsionmediated hydrothermal method. A quaternary microemulsion medium consists of Na<sub>3</sub>VO<sub>4</sub>/NaOH and

Foundation item: Projects(20601012, 20601016, 20961005) supported by the National Natural Science Foundation of China; Project(209024) supported by the Ministry of Education of China; Projects(206077, 206043, 10013-121008) supported by Inner Mongolia University, China Corresponding author: ZHANG Jun; Tel: +86-471-4992175; E-mail: cejzhang@imu.edu.cn

s232

 $RE(NO_3)_3$ aqueous surfactant solution, cetyltrimethylammonium bromide (CTAB), cosurfactant *n*-hexanol and oil phase *n*-heptane were developed, in which the microemulsion droplets acting as microreactors during the reaction process could show confinement effects allowing the formation of small size YVO<sub>4</sub>:RE nanoparticles with relatively narrow size distribution and less aggregation. The microemulsionmediated hydrothermal method shows superiority in obtaining YVO<sub>4</sub>:RE nanoparticles with controllable size, narrow size distribution and less aggregation, and may be potentially effective for synthesis of a wide series of rare earth doped luminescent nanomaterials.

#### **2** Experimental

## 2.1 Preparation of YVO<sub>4</sub>:RE (RE=Yb<sup>3+</sup>/Er<sup>3+</sup>, Yb<sup>3+</sup>/Tm<sup>3+</sup>) nanoparticles

 $RE(NO_3)_3$  (RE=Yb<sup>3+</sup>, Er<sup>3+</sup>, Tm<sup>3+</sup>), Na<sub>3</sub>VO<sub>4</sub>, hexanol, heptane, and cetyltrimethylammoniumbromide (CTAB) were purchased from J&K Chemical Company, and were used as received.

The synthesis of undoped YVO<sub>4</sub> and YVO<sub>4</sub>:RE  $(RE=Yb^{3+}/Er^{3+})$  $Yb^{3+}/Tm^{3+}$ ) nanoparticles via microemulsion-mediated hydrothermal method follows the same procedures. Taking the preparation of YVO<sub>4</sub>:Yb<sup>3+</sup>/Er<sup>3+</sup> as an example, two microemulsions were separately prepared. One contained CTAB, hexanol, heptane and aqueous solution of Na<sub>3</sub>VO<sub>4</sub>, and NaOH, and the other consisted of CTAB, hexanol, heptane and aqueous solution of  $Y(NO)_3$ ,  $Yb(NO)_3$  and  $Er(NO)_3$ (Table 1). Then one microemulsion was dropwise added into the other at room temperature with continuous stirring. Subsequently, the mixture was transferred into a 50 mL autoclave for hydrothermal treatment at 150 for 2 h to make the nanoparticles well crystallized. By

naturally cooled down to room temperature, the

 $YVO_4:Yb^{3+}/Er^{3+}$  nanoparticles were separated from the microemulsion media by centrifugation and were washed with absolute ethanol and distilled water several times. Finally, the obtained  $YVO_4:Yb^{3+}/Er^{3+}$  nanoparticles were dried at room temperature. Similar procedures were followed for the preparation of undoped  $YVO_4$  and  $YVO_4:Yb^{3+}/Tm^{3+}$  nanoparticles. For synthesis of  $YVO_4:Yb^{3+}/Tm^{3+}$  nanoparticles, the same reaction conditions were kept except using  $Tm(NO)_3$  to substitute  $Er(NO)_3$ .

For comparison, a solid annealing diffusion method was carried out to synthesize  $YVO_4$ :RE (RE=Yb<sup>3+</sup>/Er<sup>3+</sup>, Yb<sup>3+</sup>/Tm<sup>3+</sup>) nanoparticles. Typically, a certain amount of YVO<sub>4</sub> nanoparticles obtained via microemulsionmediated hydrothermal method were weighted, and a certain amount of Yb(NO<sub>3</sub>)<sub>3</sub> (20%, mole fraction) and Er(NO<sub>3</sub>)<sub>3</sub> or (Tm(NO<sub>3</sub>)<sub>3</sub> (2%, mole fraction) were weighted and mixed with YVO<sub>4</sub> nanoparticles. After grinding , the samples were calcined at 600 for 4 h to obtain the final YVO<sub>4</sub>:RE (RE=Yb<sup>3+</sup>/Er<sup>3+</sup>, Yb<sup>3+</sup>/Tm<sup>3+</sup>) nanoparticles.

#### 2.2 Characterization

Powder X-ray diffractometry (XRD) was used to characterize the phase structures of the nanoparticles. Measurements were performed using a Bruker AXS-D8 diffractometer (German) operated at 40 kV and 40 mA with a slit of 1/2 at a scanning rate of 3 (°)/min in a scanning range  $2\theta$  of 20–80°, using Cu K<sub> $\alpha$ </sub> radiation ( $\lambda$ =0.154 06 nm). Samples for XRD analysis were prepared by gently crushing the obtained products with a mortar and pestle, and were placed in a quartz glass holder for characterization. Transmission electron microscopy (TEM) characterization was performed on a JEM–2010 system operated at an acceleration voltage of 200 kV to evaluate the structure, size and shape of the nanoparticles.

Sample	Microemulsion	<i>m</i> (CTAB)/g	V(Hexanol)/mL	V(Heptane)/mL	Aqueous solution
YVO <sub>4</sub>	MA	0.5	0.9	5.6	0.3 mL Na <sub>3</sub> VO <sub>4</sub> (0.1 mol/L)+0.2 mL H <sub>2</sub> O+0.1 mL NaOH (1 mol/L), pH=10
	MB	0.5	0.9	5.6	0.3 mL Y(NO) <sub>3</sub> (0.1 mol/L)+0.3 mL H <sub>2</sub> O
YVO4:Yb <sup>3+</sup> /Er <sup>3+</sup>	MA	0.5	0.9	5.6	0.3 mL Na <sub>3</sub> VO <sub>4</sub> (0.1 mol/L)+0.2 mL H <sub>2</sub> O+0.1 mL NaOH (1 mol/L), pH=10
	MB	0.5	0.9	5.6	0.3 mL Y(NO) <sub>3</sub> (0.078 mol/L)+0.2 mL Yb(NO) <sub>3</sub> (0.03 mol/L)+0.1 mL Er(NO) <sub>3</sub> (6 mmol/L)
YVO4:Yb <sup>3+</sup> /Tm <sup>3+</sup>	MA	0.5	0.9	5.6	0.3 mL Na <sub>3</sub> VO <sub>4</sub> (0.1 mol/L)+0.2 mL H <sub>2</sub> O+0.1 mL NaOH (1 mol/L), pH=10
	MB	0.5	0.9	5.6	0.3 mL Y(NO) <sub>3</sub> (0.078 mol/L)+0.2 mL Yb(NO) <sub>3</sub> (0.03 mol/L)+0.1 mL Tm(NO) <sub>3</sub> (6 mmol/L)

Table 1 Compositions of microemulsion for preparation of YVO<sub>4</sub>, YVO<sub>4</sub>:Yb<sup>3+</sup>/Er<sup>3+</sup> and YVO<sub>4</sub>:Yb<sup>3+</sup>/Tm<sup>3+</sup> nanoparticles

Samples for TEM analysis were prepared by drying a drop of nanoparticles dispersied on an amorphous carbon-coated copper grid for the observation. Energy dispersive X-ray spectroscopy (EDX) was performed using an EDAX system attached to TEM.

#### **3 Results and discussion**

#### 3.1 X-ray powder diffractometry

The crystalline phases of YVO<sub>4</sub>. YVO<sub>4</sub>:Yb<sup>3+</sup>/Er<sup>3+</sup> and YVO<sub>4</sub>:Yb<sup>3+</sup>/Tm<sup>3+</sup> nanoparticles were determined by X-ray powder diffractometry, and XRD patterns are shown in Fig.1. The XRD patterns suggest that the good crystallinity of the nanoparticles with phase structures indexed to tetragonal structured YVO<sub>4</sub> (JCPDS No.17-0341). No impurity phase except tetragonal structured YVO4 was evidenced from the XRD characterization. Compared with bulk YVO4, the diffraction peaks were broadened, indicating the formation of nanosized YVO<sub>4</sub>. The average particles size of the nanoparticles can be roughly deduced to about 30 nm by fitting the full width at half maximum of (200) peaks according to Scherrer equation [14], which was consistent with the results determined by TEM observations.



Fig.1 XRD patterns of  $YVO_4$ : $Yb^{3+}/Er^{3+}(a)$  and  $YVO_4$ : $Yb^{3+}/Tm^{3+}(b)$  nanoparticles

#### 3.2 TEM characterization

The sizes and shapes of the as-prepared YVO<sub>4</sub>,  $YVO_4$ :Yb<sup>3+</sup>/Er<sup>3+</sup> and  $YVO_4$ :Yb<sup>3+</sup>/Tm<sup>3+</sup> nanoparticles prepared via microemulsion-mediated hydrothermal method were examined by TEM, and the images are shown in Fig.2. Fig.2(a) displays the TEM image of  $YVO_4$  nanoparticles without doping. It can be seen from



**Fig.2** TEM images of  $YVO_4(a)$  and (b),  $VO_4:Yb^{3+}/Er^{3+}(c)$ , and  $YVO_4:Yb^{3+}/Tm^{3+}(d)$  nanoparticles prepared via microemulsion-mediated hydrothermal method

Fig.2(a) that the nanoparticles show the spherical shape with particle sizes of about 30 nm. Fig.2(b) presents the high resolution TEM image of YVO<sub>4</sub> nanoparticles. The lattice fringe observed from the HRTEM image suggests the formation of well crystalline YVO<sub>4</sub> nanoparticles with tetragonal phase structure in consistence with XRD results. Fig.2(c) and (d) show the TEM images of YVO<sub>4</sub>:Yb<sup>3+</sup>/Er<sup>3+</sup> and YVO<sub>4</sub>:Yb<sup>3+</sup>/Tm<sup>3+</sup> nanoparticles, respectively. It is found that after co-doping the Yb<sup>3+</sup>/Er<sup>3+</sup> and Yb<sup>3+</sup>/Tm<sup>3+</sup> ions entered into YVO<sub>4</sub>, no changes of phase structure of YVO<sub>4</sub> nanoparticles were observed, and the particle shape and size keep almost the same, indicating the effectiveness of microemulsion method for homogenous doping of Yb<sup>3+</sup>/Er<sup>3+</sup> and Yb<sup>3+</sup>/Tm<sup>3+</sup> ions into YVO<sub>4</sub>.

#### 3.3 Energy dispersive X-ray analysis

To confirm the contents of Yb, Er and Tm elements existing in the doped YVO<sub>4</sub> nanoparticles, energy dispersive X-ray (EDX) analysis was carried out, as shown in Fig.3. Beside V, O and Y elements, the representative peaks of Yb/Er and Yb/Tm elements appear in the spectra of YVO<sub>4</sub>:Yb<sup>3+</sup>/Er<sup>3+</sup> and YVO<sub>4</sub>:Yb<sup>3+</sup>/Tm<sup>3+</sup> nanoparticles, respectively, which confirms the existence of Yb/Er and Yb/Tm elements in YVO<sub>4</sub> nanoparticles and suggests the realization of the



**Fig.3** EDX patterns of as-prepared  $YVO_4$ :  $Yb^{3+}/Er^{3+}(a)$  and  $YVO_4$ :  $Yb^{3+}/Tm^{3+}(b)$  nanoparticles

co-doping of  $Yb^{3+}/Er^{3+}$  and  $Yb^{3+}/Tm^{3+}$  ions into  $YVO_4$  nanoparticles via microemulsion mediated hydrothermal method.

# 3.4 Comparison of different preparation methods on effect of size and shape of YVO<sub>4</sub>:Yb<sup>3+</sup>/Er<sup>3+</sup> and YVO<sub>4</sub>:Yb<sup>3+</sup>/Tm<sup>3+</sup> nanoparticles

To compare the co-doping effectiveness, the particle sizes and shapes of nanoparticles prepared via microemulsion mediated hydrothermal method were compared with those by other methods, a solid annealing diffusion approach, as described in experimental section. The size and shape of the nanoparticles obtained by solid annealing diffusion method are shown in Fig.4. It is found that solid annealing diffusion method may result in the formation of YVO<sub>4</sub>:Yb<sup>3+</sup>/Er<sup>3+</sup> and YVO<sub>4</sub>:Yb<sup>3+</sup>/Tm<sup>3+</sup> nanoparticles with larger particle size and severe aggregation. Meanwhile, the XRD characterization of the YVO<sub>4</sub>:Yb<sup>3+</sup>/Er<sup>3+</sup> and YVO<sub>4</sub>:Yb<sup>3+</sup>/Tm<sup>3+</sup> obtained nanoparticles indicates the formation of Y2O3 phase except the tetragonal structured YVO<sub>4</sub>. This suggests that the microemulsion-mediated hydrothermal method shows



Fig.4 TEM images of  $\rm YVO_4{:}Yb^{3+}/\rm Er^{3+}(a)$  and  $\rm YVO_4{:}Yb^{3+}/\rm Tm^{3+}(b)$  nanoparticles prepared via solid annealing diffusion method

superiority in obtaining YVO<sub>4</sub>:Yb<sup>3+</sup>/Er<sup>3+</sup> and YVO<sub>4</sub>:Yb<sup>3+</sup>/Tm<sup>3+</sup> nanoparticles with controllable size, narrow size distribution and less aggregation.

## **4** Conclusions

1) Microemulsion-mediated hydrothermal method was successfully developed for preparation of YVO<sub>4</sub>, YVO<sub>4</sub>:Yb<sup>3+</sup>/Er<sup>3+</sup> and YVO<sub>4</sub>:Yb<sup>3+</sup>/Tm<sup>3+</sup> nanoparticles.

2) The microemulsion-mediated hydrothermal method shows effectiveness in obtaining  $YVO_4$ ,  $YVO_4$ : $Yb^{3+}/Er^{3+}$  and  $YVO_4$ : $Yb^{3+}/Tm^{3+}$  nanoparticles with small size and narrow size distribution and less aggregation.

3) Compared with the conventional solid annealing diffusion method, homogenous co-doping of  $Yb^{3+}/Er^{3+}$  and  $Yb^{3+}/Tm^{3+}$  into  $YVO_4$  lattice can be easily realized by the microemulsion-mediated hydrothermal method.

4) The microemulsion-mediated hydrothermal method may be potentially applicable for synthesis of other rare earth doped up-converting luminescence nanomaterials.

## References

- LEVINE A K, PALILLA F C. A new, highly efficient red-emitting cathodoluminescent phosphor (YVO4:Eu) for color television [J]. Appl Phy Lett, 1964, 5: 118–124.
- [2] ZHANG M, SHI S, MENG J, WANG X, FAN H, ZHU Y, WANG X, QIAN Y. Preparation and characterization of near-infrared luminescent bifunctional core/shell nanocomposites [J]. J Phys Chem C, 2008, 112: 2825–2830.

- [3] KRÄMER K W, BINER D, FREI G, GÜDEL H U, HEHLEN M P, LÜTHI S R. Hexagonal sodium yttrium fluoride based green and blue emitting upconversion phosphors [J]. Chem Mater, 2004, 16: 1244–1251.
- [4] BOYER J C, CUCCIA L A, CAPOBIANCO J A. Synthesis of colloidal upconverting NaYF4:Er<sup>3+</sup>/Yb<sup>3+</sup> and Tm<sup>3+</sup>/Yb<sup>3+</sup> monodisperse nanocrystals [J]. Nano Letters, 2007, 7: 847–852.
- [5] SUYER J F, GRIMM J, KRÄMER K W, GÜDEL H U. Highly efficient near-infrared to visible up-conversion process in NaYF<sub>4</sub>:Er<sup>3+</sup>, Yb<sup>3+</sup> [J]. J Lumin, 2005, 114: 53–59.
- [6] CHEN Z G, CHEN H L, HU H, YU M X, LI F Y, ZHANG Q, ZHOU Z G, YI T, HUANG C H. Versatile synthesis strategy for carboxylic acid- functionalized upconverting nanophosphors as biological labels [J]. J Am Chem Soc, 2008, 130: 3023–3029.
- [7] YANG K, ZHENG F, WU R, LI H, ZHANG X. Upconversion luminescent properties of YVO<sub>4</sub>:Yb<sup>3+</sup>, Er<sup>3+</sup> nano-powder by sol-gol method [J]. J Rare Earths, 2006, 24: 162–166.
- [8] CHEN X, LIU K, ZHUANG J, WANG G, CHEN C. The upconversion luminescent research of HoYb:YVO<sub>4</sub> [J]. Acta Physica Sinica, 2002, 51: 690–695.
- [9] ZHANG H, FU X, NIU S, SUN G, XIN Q. Photoluminescence of nanocrystalline YVO<sub>4</sub>:Tm<sub>x</sub>Dy<sub>1-x</sub> prepared by a modified pechini method [J]. Materials Letters, 2007, 61: 308–311.
- [10] SUN Y, LIU H, WANG X, KONG X, ZHANG H. Optical spectroscopy and visible upconversion studies of YVO<sub>4</sub>:Er<sup>3+</sup> nanocrystals synthesized by a hydrothermal process [J]. Chem Mater, 2006, 18: 2726–2732.
- [11] PENG H, HUANG S, SUN L, YAN C. Analysis of surface effect on luminescent properties of Eu<sup>3+</sup> in YVO<sub>4</sub> nanocrystals [J]. Physics Letters A, 2007, 367: 211–214.
- [12] ZHANG J, SUN L, LIAO C, YAN C. Size control and photoluminescence enhancement of CdS nanoparticles prepared via reverse micelle method [J]. Solid State Commun, 2002, 124: 45–48.
- [13] SUN L, ZHANG Y, ZHANG J, YAN C, LIAO C, LU Y. Fabrication of size controllable YVO<sub>4</sub> nanoparticles via microemulsion-mediated synthetic process [J]. Solid State Commun, 2002, 124: 35–38.
- [14] WARREN B E. X-ray diffraction[M]. New York: Dover Publications Inc, 1990.

#### (Edited by CHEN Ai-hua)